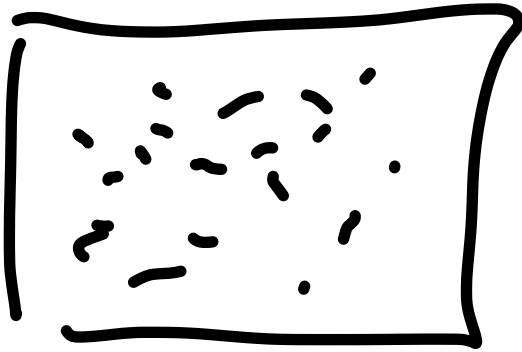


Lecture 6 - Microcanonical to Canonical

Previously:

N, V, E :



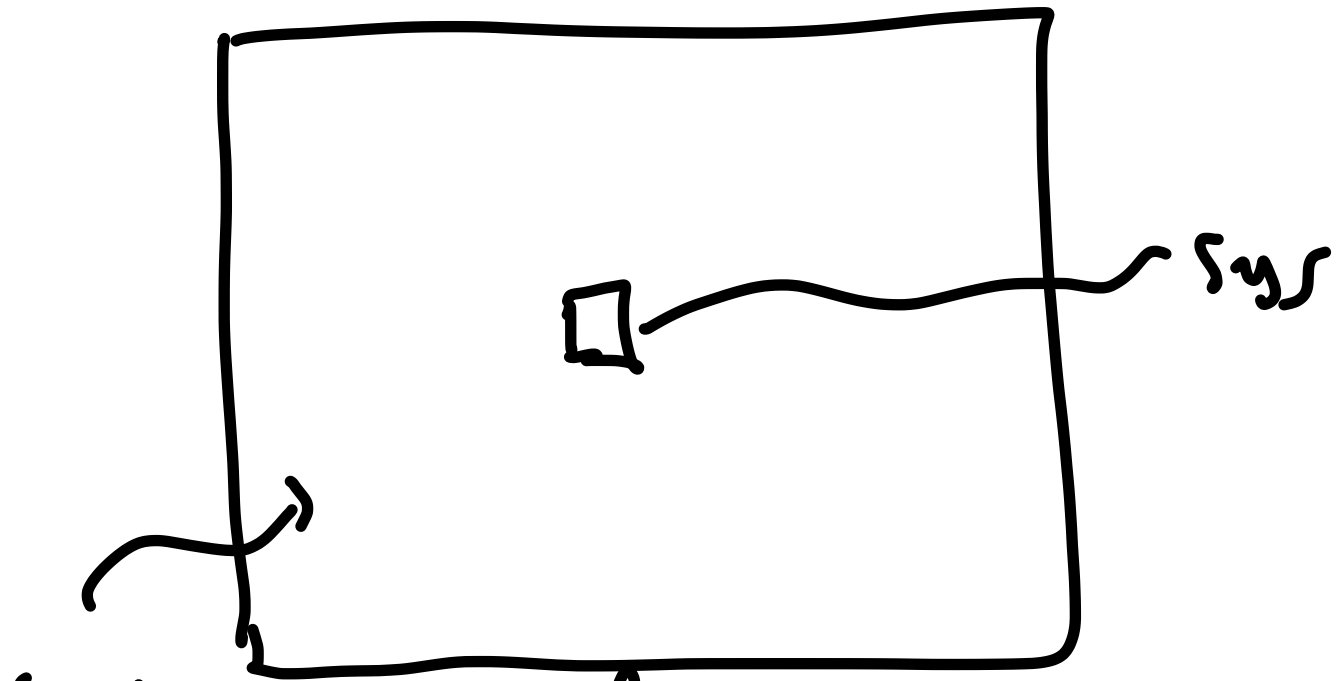
$$P(\vec{x}) \propto \begin{cases} \frac{1}{\Omega(N, V, E)} & \text{if } \mathcal{H}(x) = E \\ 0 & \text{otherwise} \end{cases}$$

$$S = k_B \ln \Omega(N, V, E)$$

N, V, T :

$$P(\vec{x}) \propto e^{-\mathcal{H}(x)/k_B T}$$

but why?



bath of Plows until $1/T$ equal

$$\text{Total: } \Omega = \Omega_{\text{sys}} \Omega_{\text{bath}}$$

$$\Omega_{\text{bath}}(N, V, E_{\text{bath}})$$

$$E_{\text{tot}} = E_{\text{bath}} + E_{\text{sys}} \text{ is const}$$

Suppose system has config \vec{X}

$$E_{\text{sys}} = \mathcal{H}(X) \Rightarrow E_{\text{bath}} = E - E_{\text{sys}} \text{ fixed}$$

$$P(\vec{X}) \propto \Omega_{\text{b}}(N, V, E_{\text{b}})$$

$$S(N, V, E_b) \approx S(N, V, E_{tot})$$

$$+ (E_b - E_{tot}) \frac{dS}{dE_b} + \frac{1}{2} (E_b - E_{tot})^2 \frac{\partial^2 S}{\partial E_b^2} + \dots$$

$$S(N, V, E_b) \approx \text{const} - E_{sys} \cdot \frac{1}{T_{bath}} = \frac{1}{T}$$

$$k_B \ln \Omega(N, V, E_b)$$

$$\Omega(N, V, E_b) \propto e^{-E_{sys} \cdot \frac{1}{k_B T}}$$

$$\text{so } P(x) \propto e^{-\mathcal{H}(x)/k_B T} \quad \checkmark$$

$$P(x) = \frac{1}{Z} e^{-\mathcal{H}(x)/k_B T}$$

$$Z = \int d\vec{x} e^{-\beta \mathcal{H}(x)} \quad \beta = \frac{1}{k_B T}$$

often, define

$$Q = \frac{1}{h^{3N} N!} \int d\vec{x} e^{-\beta \mathcal{H}(x)}$$

$$P(x) = \frac{1}{h^{3N} N!} e^{-\beta \mathcal{H}(x)} / Q(N, V, T)$$
$$= e^{-\beta \mathcal{H}(x)} / Z(N, V, T)$$

Can write Q in another useful way:

$$Q = \frac{1}{h^{3N} N!} \int d\vec{x} e^{-\beta \mathcal{H}(x)}$$

$$= \frac{1}{h^{3N} N!} \int d\vec{x} \int_0^{\infty} d\varepsilon \delta(\mathcal{H}(x) - \varepsilon) e^{-\beta \varepsilon}$$

$$= \int_0^{\infty} d\varepsilon \left[\frac{1}{h^{3N} N!} \int dx \delta(\mathcal{H}(x) - \varepsilon) \right] e^{-\beta \varepsilon}$$

$$\frac{1}{\varepsilon_0} \Omega(N, V, \varepsilon) \leftarrow \text{system}$$

$$= \frac{1}{\varepsilon_0} \int_0^{\infty} d\varepsilon \Omega(N, V, \varepsilon) e^{-\beta \varepsilon}$$

$\underbrace{\hspace{2cm}}$
 degeneracy
 density of states $\omega(\varepsilon)$

[Laplace transform]

$$P(\varepsilon)_{\text{sys}} \propto \omega(\varepsilon) e^{-\beta \varepsilon}$$

More thermodynamics

To derive more, need to know how ensembles are connected in classical thermo

Microcanonical:

$S(N, V, E)$ is fundamental, $\max_{E, V}$

$$dS = \left(\frac{\partial S}{\partial N} \right) dN + \left(\frac{\partial S}{\partial V} \right) dV + \left(\frac{\partial S}{\partial E} \right) dE$$

$$= \frac{1}{T} dE + \frac{P}{T} dV - \frac{\mu}{T} dN$$

at same time, 1st law

$$dE = T dS - P dV + \mu dN$$

here $E(N, V, S)$ but can't measure S

In math, can replace an argument by a conjugate variable through Legendre transform to get a new function

$$A(N, V, T) = E(N, V, S) - S \left(\frac{\partial E}{\partial S} \right)_{N, V}$$

$$= E - TS$$

↑ Helmholtz free energy

$$dA = \left(\frac{\partial A}{\partial N} \right) dN + \left(\frac{\partial A}{\partial V} \right) dV + \left(\frac{\partial A}{\partial T} \right) dT$$

$$= dE - d[TS] = dE - TdS - SdT$$

$$= -PdV + \mu dN - SdT$$

$$\left(\frac{\partial A}{\partial N} \right)_{V, T} = \mu \quad \left(\frac{\partial A}{\partial T} \right)_{N, V} = -S \quad \left(\frac{\partial A}{\partial V} \right)_{N, T} = -P$$

A is thermodynamic potential
for N, V, T , minimized @ eq

Now will show $A = -k_B T \ln z$
like entropy $\longrightarrow = -\frac{1}{\beta} \ln z$

$$A = \mathcal{E} - TS = \mathcal{E} + T \left(\frac{\partial A}{\partial T} \right)_{N, V}$$

$$\frac{\partial A}{\partial T} = \frac{\partial A}{\partial \beta} \cdot \frac{\partial \beta}{\partial T} = -\frac{1}{k_B T^2} \frac{\partial A}{\partial \beta}$$

$$\text{so } T \frac{\partial A}{\partial T} = -\beta \frac{\partial A}{\partial \beta}$$

already showed $\langle \mathcal{E} \rangle = -\frac{\partial \ln z}{\partial \beta}$

$$A = -\frac{\partial \ln z}{\partial \beta} - \beta \frac{\partial A}{\partial \beta} \quad \text{or} \quad \left[\frac{1}{\beta^2} \ln z - \frac{1}{\beta} \frac{\partial \ln z}{\partial \beta} \right]$$

$$= -\frac{1}{\beta} \ln z \quad \checkmark$$

$$\text{So } A = -k_B T \ln Z = -k_B T \ln Q + \text{const}$$

Consider ideal gas

$$Q = \frac{1}{N! h^{3N}} \int d\vec{x} \int d\vec{p} e^{-\beta \sum p_i^2 / 2m}$$

$$= \frac{1}{N! h^{3N}} V^N \cdot (2\pi m k_B T)^{3/2 N}$$

$$= \frac{1}{N!} V^N \left(\frac{2\pi m k_B T}{h^2} \right)^{3/2 N}$$

$$= \frac{1}{N!} \left(\frac{V}{\Lambda^3} \right)^N$$

$$\mathcal{E} = - \frac{\partial \ln Q}{\partial \beta} = - \frac{\partial}{\partial \beta} [\ln(\beta^{-3/2 N})]$$

$$= 3/2 N k_B T \quad \checkmark$$

$$P = - \frac{\partial A}{\partial V} = k_B T \frac{\partial \ln Q}{\partial V} \Rightarrow P V = N k_B T \quad \checkmark$$