

# Lecture 5 - Ideal Gas

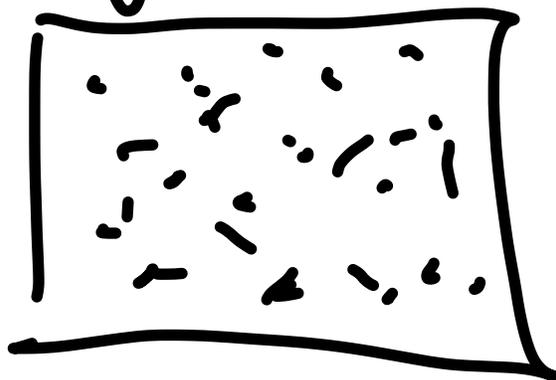
Last time:

Entropy: state function  
which does not decrease in  
changing the state of a system

For isolated system, const  $N, V, E$ :

$$\frac{1}{T} = \left( \frac{\partial S}{\partial E} \right)_{N, V} \quad - \frac{P}{T} = \left( \frac{\partial S}{\partial V} \right)_{N, E}$$

Now let's consider the properties  
of a gas: generally:  $p = N/V kT$



molecules don't  
"feel" each other

Can say approximately,  

$$U(\vec{x}) = \begin{cases} 0 & \text{if } x_i \in (0, L) \forall i \\ \infty & \text{otherwise} \end{cases}$$

$$H(\vec{x}) = \sum_{i=1}^N \vec{p}_i^2 / 2m + \begin{cases} \infty & \text{x outside box} \\ 0 & \text{otherwise} \end{cases}$$

$$\Omega = \frac{\epsilon_0}{h^{3N} N!} \int d^3 p^{3N} \delta(\sum \frac{p_i^2}{2m} - \epsilon) \int d^3 x^{3N}$$

$$\Omega = \frac{\epsilon_0}{h^{3N} N!} V^N \int d^3 p^{3N} \delta(\sum \frac{p_i^2}{2m} - \epsilon)$$

What is this prefactor:

$\epsilon_0$ :  $\delta(x)$  has units  $\frac{1}{x}$

otherwise  $\int dx \delta(x) = 1$  would have units

$N!$ : how many ways relabel

$N$  particles?  $N(N-1) \dots$

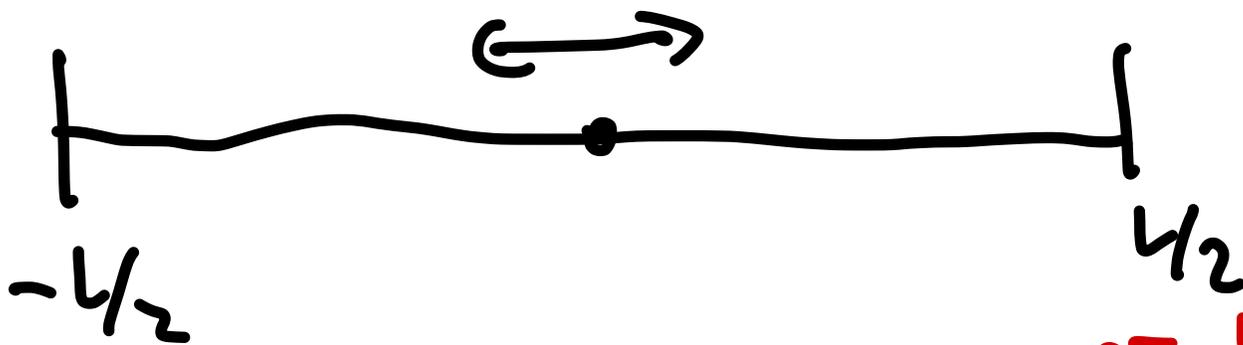
These arrangements indistinguishable

$\{ \epsilon_0, N_A!, N_B!, N_C!, \dots \}$

Without  $N!$ , will get a paradox (later)

- $h^{3N}$ :
- cancels units of  $dx \cdot dp$
  - hard way: enumerate all states, take all possible  $x$  &  $p$ , cannot know better than  $\Delta x \Delta p \sim h$
  - final result will connect to another QM concept

To see strategy, 16 first:



Tuckman 3.5

$$S = k_B \ln \Omega(L, L, \epsilon)$$

$$\Omega = \frac{\epsilon_0}{h} \int_{-L}^L dp \delta\left(\frac{p^2}{2m} - \epsilon\right) \int_{-L/2}^{L/2} dx$$

$$\Omega_{1d} = \frac{L \epsilon_0}{h} \int_{-\infty}^{\infty} dp \delta(p^2/2m - \epsilon)$$

$$\left[ \begin{array}{l} y^2 = p^2/2m \\ dy = \frac{1}{\sqrt{2m}} dp \end{array} \right.$$

$$= \frac{L \epsilon_0}{h} \sqrt{2m} \int_{-\infty}^{\infty} dy \delta(y^2 - \epsilon)$$

$$\left[ \delta(y^2 - \epsilon) = (\delta(y - \sqrt{\epsilon}) + \delta(y + \sqrt{\epsilon})) \cdot \frac{1}{2\sqrt{\epsilon}} \right.$$

$$= \epsilon_0 \frac{L}{h} \sqrt{\frac{2m}{\epsilon}} \quad \checkmark$$

Now what happens for  $3N$ ?

$$\Omega(N, V, \epsilon) = \frac{\epsilon_0}{h^{3N} N!} V^N \int d\mathbf{p}^{3N} \delta\left(\sum_{i=1}^N p_i^2 - \epsilon\right)$$

first sub  $y_i^2 = p_i^2 / 2m$

$$= \frac{\epsilon_0}{h^{3N} N!} V^N \cdot (2m)^{3N/2} \int d\mathbf{y}^{3N} \delta\left(\sum_{i=1}^N y_i^2 - \epsilon\right)$$

if 3d and we had  $x^2 + y^2 + z^2 = r^2$

$$\text{Jacobian} = \int d\theta d\phi r^2 \sin\theta dr d\theta d\phi = \underbrace{4\pi r^2 dr}$$

Here,  $3N$  dimensional version of

Sphere  $S_{n-1}$

$$d\mathbf{y}^{3N} = r^{3N-1} S_{3N-1} dr$$

In HW, rest of details filled in

final result:

$$\Omega \approx \frac{\epsilon_0}{N!} \left( \frac{V}{h^3} \left( \frac{4\pi m \epsilon_0}{3N} \right)^{3/2} \right)^N$$

↙ large N

↗ note: depends on  $N, V, \epsilon,$  and also  $m,$  but derivs will not cont these

$$S = k_B \ln \Omega$$

$$\frac{1}{k_B T} = \frac{\partial \ln \Omega}{\partial \epsilon} = \frac{3}{2} N \cdot \frac{1}{\epsilon}$$

$$\Rightarrow \epsilon = \frac{3}{2} N k_B T = \frac{3}{2} N R T \quad \checkmark$$

$$\frac{P}{T} = + k_B \frac{\partial \ln \Omega}{\partial V} = N k_B \cdot \frac{1}{V}$$

$$P V = N k_B T = N R T \quad \checkmark$$

$$\Omega \approx \frac{\epsilon_0}{N!} \left( \frac{V}{h^3} \left( \frac{4\pi m \epsilon_0}{3N} \right)^{3/2} \right)^N \quad \text{--- } 3/2 N k_B$$

[approx messes up units, but still ok]

$$\approx \frac{\epsilon_0}{N!} \left[ V \cdot \left( \frac{2\pi m k_B T}{h^2} \right)^{3/2} \right]^N \cdot e^{3N/2}$$

$$\Lambda = \sqrt{\frac{h^2}{2\pi m k_B T}} \quad \text{--- thermal wavelength}$$

$\Omega_1$  particle  $\propto \frac{V}{\Lambda^3}$  ~ how many boxes

$$\Omega_N \approx \frac{1}{N!} \Omega_1^N \quad \sim N \ln N - N$$

$$S = k_B \ln \Omega \approx N k_B \ln \left[ \frac{V}{\Lambda^3} \right] - k_B \ln N! + \frac{3}{2} N k_B + C$$

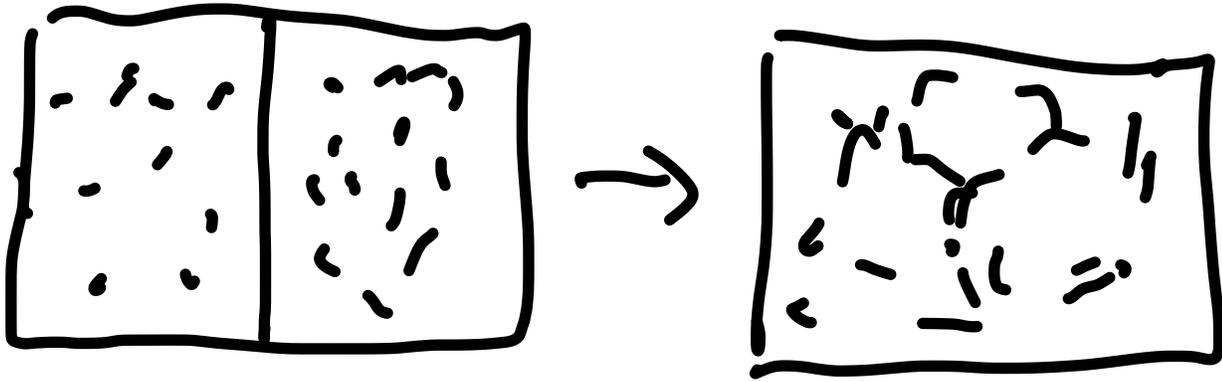
$$= N k_B \ln \left[ \frac{1}{N} \frac{V}{\Lambda^3} \right] + \frac{5}{2} N k_B + C$$

$$= N k_B \ln \left[ \frac{V}{N} \left( \frac{2\pi m k_B T}{h^2} \right)^{3/2} \right] + \frac{5}{2} N k_B + \text{const}$$

Sackur-Tetrode equation

$V/N\Lambda^3 \gg 1$ , classical regime is where valid

Why do we need  $N!$ ? Gibbs paradox



Remove wall

What is  $S_2 - 2S_1$ ?

Compute with & without  $N!$   
factor (see HW)