

Lecture 5 -

Canonical Ensemble

Other thermodynamic ensembles

Idea: Our hypothesis before was that if we could enumerate all the states of a system and the likelihood of seeing them, then we could predict any observable:

$$\langle A \rangle = \sum_{\text{states}} P(n) A(n)$$

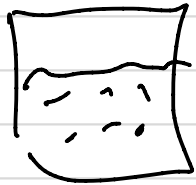
For closed isolated system, const N, V, E
we expect all states to have the same weight, $\Omega(N, V, E)$ states

$$\langle A \rangle = \int d\vec{q} d\vec{p} \delta(H(\vec{q}, \vec{p}) - E) A(\vec{q}, \vec{p}) / \int d\vec{q} d\vec{p} \delta(H(\vec{q}, \vec{p}) - E)$$

But we don't live in an isolated closed system \rightarrow

Chemistry usually happens at const

$T \& V$



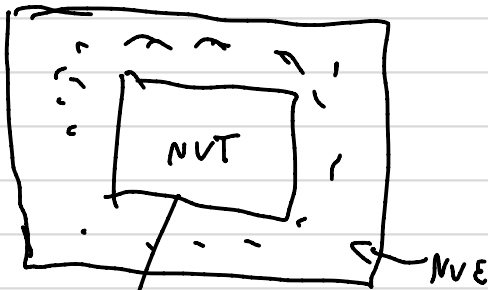
or

$T \& P$



So we have to see how likely a configuration is under these conditions

First, how is this dealt with in classical thermo



Impervious to particles, rigid, can exchange heat



can exchange heat, change volume (can show pressure equilibrate @ eq)

(can also have permeable membrane, eg μ, V, T)

In microcanonical ensemble we had $S(N, V, E)$ &

$$dS = \frac{1}{T} dE + \frac{P}{T} dV - \frac{\mu}{T} dN$$

lets rewrite in terms of E , b/c that may be easier to think about

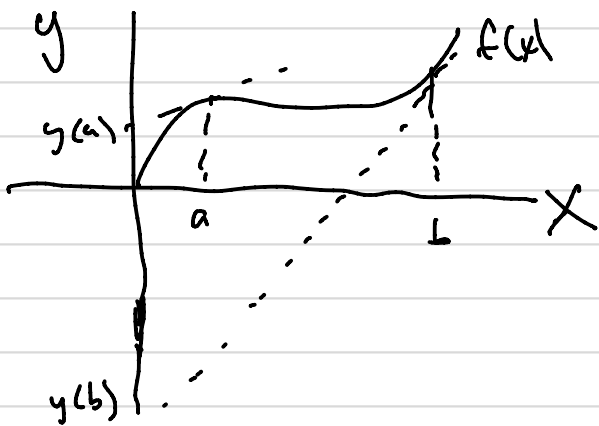
$$dE = T dS - P dV + \mu dN$$

This means $T = \left(\frac{\partial E}{\partial S}\right)_{N,P}$ $P = -\left(\frac{\partial E}{\partial V}\right)_{S,N}$ $\mu = \left(\frac{\partial E}{\partial N}\right)_{S,V}$

And $E(N, V, S)$ is a state function.

T , $-P$, and μ are called conjugate variables of S, V & N respectively

If we want a thermodynamic state function that depends on a conjugate variable instead of a current variable we have a trick called a Legendre transform



The conj of x is u
 $u = \left(\frac{\partial f}{\partial x}\right) = f'(x)$
 $x = (f')^{-1}(u)$
 $= \phi(a)$
 Slope intercept says

$$f(a) = \underbrace{f'(a)}_{u(a)} a + y(a)$$

$$f(b) = u(b)b + y(b)$$

in general $f(x) = u(x)x + y(x)$

$$y(x) = f(x) - u(x)x$$

new func $\rightarrow \tilde{f}(u) = f(x(u)) - x(u) \frac{\partial f}{\partial x}$

Eg. New func $A(N, V, T) = E(N, V, S) - S \left(\frac{\partial E}{\partial S}\right)_{N, V} = \underline{\underline{E(N, V, S) - TS}}$

\uparrow \uparrow
 μ u above x above

A is the "Helmholtz free energy"

Chain rule for $A(N, U, T)$ gives

$$dA = \left(\frac{\partial A}{\partial T} \right)_{N, U} dT + \left(\frac{\partial A}{\partial U} \right)_{T, N} dU + \left(\frac{\partial A}{\partial N} \right)_{U, T} dN$$

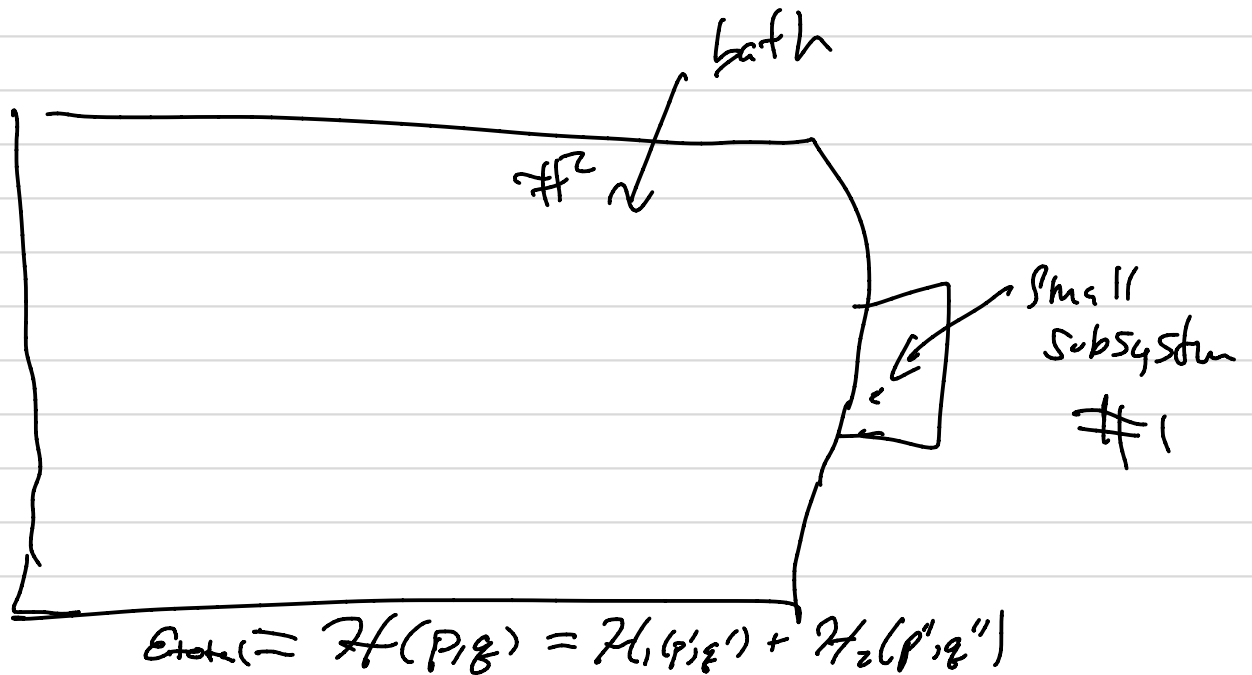
$$\begin{aligned} \text{but } A = E - TS \Rightarrow dA &= dE - Tds - SdT \\ &= (Tds - PdV + \mu dN) - Tds - SdT \\ &= -SdT - PdV + \mu dN \end{aligned}$$

$$\Rightarrow \left(\frac{\partial A}{\partial T} \right)_{N, U} = -S \quad \left(\frac{\partial A}{\partial U} \right)_{N, T} = -P \quad \left(\frac{\partial A}{\partial N} \right)_{U, T} = \mu$$

*

$\underbrace{\hspace{10em}}$
A is like an energy

This is all classical thermo. To derive things based on a particle basis, we need to figure out how likely a state is, i.e. the inside of the partition function



Integrate over coordinates (p'', q'') , this is a type of what we call "coarse-graining", reduce complexity of system. May cover this more later

What is $f(p', q')$ prob of state in subsystem.

$$f(p', q') = \frac{\int dp'' dq'' \delta(H(p'', q'') - H_1(p', q') - H_2(p'', q''))}{\Omega(N, U, \epsilon) \leftarrow \text{const}}$$

Since system 1 is small, expand function around $H_1(p', q') = 0$, convenient to expand $\ln(f)$

Remember $\phi(x) = \phi(a) + (x-a)\phi'(a) + \frac{(x-a)^2}{2}\phi''(a) + \dots$
 $= \sum_{n=0}^{\infty} \frac{\phi^{(n)}(a)}{n!} (x-a)^n$ Taylor series

$$\log f(p', q') = \log \int dp'' dq'' \delta(H(p'', q'') - H_1(p', q') - H_2(p'', q'')) - \text{const}$$

$$\approx \log \int dp'' dq'' \delta(H(p'', q'') - H_2(p'', q'')) + \left(\frac{\partial}{\partial H_1} \log \int dp'' dq'' \delta(H(p'', q'') - H_1(p', q') - H_2(p'', q'')) \right) H_1 + \dots$$

first order large system $H_1=0$

lets start $\mathcal{H}(p, q) = E$ and use property of delta func

$$\frac{\partial}{\partial H_1} \delta(H_1 + H_2 - E) = - \frac{\partial}{\partial E} \delta(H_1 + H_2 - E)$$

$$\text{so } \log f \approx \log \int dp'' dq'' \delta(H_2(p'', q'') - E)$$

$$- \left(\frac{\partial}{\partial E} \log \int dp'' dq'' \delta(H_1 + H_2 - E) \right) \Big|_{H_1=0} \mathcal{H}_1(p', q')$$

$$\left(\frac{\partial}{\partial E} \log \delta(H_2 - E) \right) \mathcal{H}_1(p', q')$$

$$\log \int dp'' dq'' \delta(H_2(p'', q'') - E) = S_2(N_2, V_2, E_2)$$

$$= S_2/k$$

$$\log f \approx \frac{S_2}{k} - \mathcal{H}_1(p', q') \frac{\partial}{\partial E} \frac{S_2}{k} (N_2, V_2, E_2)$$

$$\log f \approx \underbrace{\frac{S_2}{k}}_{\text{const}} - \mathcal{H}_1(p', q') / kT_2 \quad T_2 \approx T$$

$$\text{so } f \propto \exp(-\mathcal{H}_1(p', q') / kT) \quad \checkmark$$

The normalized prob of seeing a state is

$$f(p, q) = \frac{e^{-\mathcal{H}(p, q) / k_B T}}{\int dp \int dq e^{-\mathcal{H}(p, q) / k_B T}}$$

we often call denominator Z

and this is true even for systems where we're not thinking about particle systems, so

$$Z = \int d\vec{x} e^{-\beta H(x)} \quad \text{where } \beta = 1/k_B T$$

or for discrete systems $Z = \sum_n e^{-\beta E_n}$

for systems of indistinguishable particles, usually write

$$Q(N, V, T) = \frac{1}{N! h^{3N}} \int dp^{3N} dq^{3N} e^{-\beta H(p, q)}$$

$$f(q, p) = \frac{\frac{1}{N! h^{3N}} e^{-\beta H(q, p)}}{Q(N, V, T)} \quad \left(\frac{1}{N! h^{3N}} \text{ cancel out to give same prob. func.} \right)$$

What is the connection between Q & $\Omega(N, V, E)$?
insert delta func

$$\begin{aligned} Q(N, V, T) &= \frac{1}{N! h^{3N}} \int dp^{3N} dq^{3N} \int_0^\infty dE \delta(H - E) e^{-\beta E} \\ &= \int_0^\infty dE e^{-\beta E} \underbrace{\frac{1}{N! h^{3N}} \int dp^{3N} dq^{3N} \delta(H(p, q) - E)}_{\frac{1}{E_0} \Omega(N, V, E)} \end{aligned}$$

$$Q(N, V, T) = \frac{1}{\epsilon_0} \int_0^{\infty} dE e^{-\beta E} \Omega(N, V, E) \quad \leftarrow P(E)$$

so the canonical partition function is averaging the number of states at each energy if they have weight $e^{-\beta E}$

Discrete, $Q(N, V, T) = \sum_E \Omega(N, V, E) e^{-E/k_B T}$