

Lecture 3:

Classical Mechanics and
Microcanonical Ensemble

Last time - Hamilton's Eqs of Motion

$$\dot{q}_i = \frac{\partial \mathcal{H}}{\partial p_i} \quad \dot{p}_i = -\frac{\partial \mathcal{H}}{\partial q_i}$$

Phase space is the coordinates describing everything about the system

$$\text{So, } X(t) = \{q_1(t), q_2(t), \dots, q_N(t), p_1(t), \dots, p_N(t)\}$$

$\mathcal{H}(x)$ is one function of x , and we showed that $\frac{d\mathcal{H}(x)}{dt} = 0$ if the system follows hamiltonian/newtonian dynamics

How do other quantities change with time? By the chain rule formula from last time:

$$d a(x) / dt = \sum_{i=1}^{2N} \left(\frac{\partial a}{\partial q_i} \frac{\partial q_i}{\partial t} + \frac{\partial a}{\partial p_i} \frac{\partial p_i}{\partial t} \right) = \sum_{i=1}^{2N} \frac{\partial a}{\partial q_i} \frac{\partial \mathcal{H}}{\partial p_i} - \frac{\partial a}{\partial p_i} \frac{\partial \mathcal{H}}{\partial q_i}$$

$$\text{if we define } \{a, b\} \equiv \sum_{i=1}^{2N} \frac{\partial a}{\partial q_i} \frac{\partial b}{\partial p_i} - \frac{\partial a}{\partial p_i} \frac{\partial b}{\partial q_i} \quad (\text{Poisson Bracket})$$

$$\text{then we see that } \boxed{\frac{d a(x)}{dt} = \{a, \mathcal{H}\}}$$

A conserved quantity, one that doesn't depend on time i.e.

$$d a / dt = 0, \text{ e.g. we showed } \frac{d \mathcal{H}}{dt} = 0$$

$$\text{so } \{ \mathcal{H}, \mathcal{H} \} = 0.$$

Another example, $a(\vec{x}) = \vec{p}_{\text{tot}} = \sum_{i=1}^N \vec{p}_i$

$$\frac{d p_i}{dt} = \{ p_i, \mathcal{H} \} = -\frac{\partial \mathcal{H}}{\partial q_i} = F_i$$

$\frac{d \vec{p}_{\text{tot}}}{dt} = \sum \vec{F}_i$, so if net force is 0, momentum is conserved

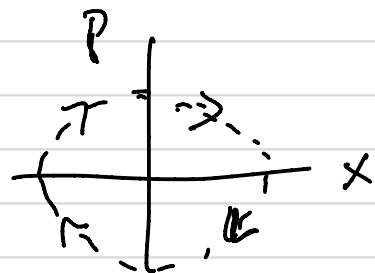
A "microstate" of a system is one particular point in phase space $X(t)$ for our system

For a conservative system, $H(x(t)) = E, \text{const}$

So the system must remain on some const E hypersurface.

Eg, if $H(x, p) = x^2/2 + p^2/2$

radius of this circle depends on initial total energy



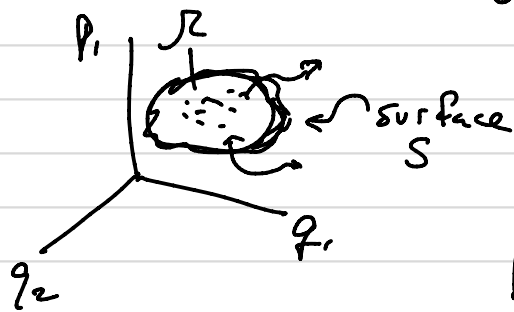
An "ensemble" is a collection of microstates all with the same macroscopic characteristics

We can express the fraction of phase space points in the ensemble within a small volume $d\vec{x}$ of point \vec{x} by

the function $f(x, t) dx$
 $\rightarrow f(x, t) \geq 0$ and $\int dx f(x, t) = 1$

We can assert that the total number of members of the ensemble stays fixed so that f stays normalized

If we draw a volume in phase space
 no sources or sinks, # points in volume =
 flux through surface



$$\text{fraction in } \Omega = \int_{\Omega} dx f(x, t)$$

can see how f changes
 by equating flux out with
 $d\Omega/dt$

This turns out to reduce to (see Chapter 2.5)

$$\frac{\partial f(x(t), t)}{\partial t} + \underbrace{\frac{dx(t)}{dt} \cdot \nabla f(x(t), t)}_{\text{flux out}} = 0 = \frac{df}{dt}$$

$$+ \sum_{i=1}^{3N} \left(\dot{q}_i \frac{\partial f}{\partial q_i} + \dot{p}_i \frac{\partial f}{\partial p_i} \right)$$

$\uparrow \frac{\partial H}{\partial p_i} \quad \quad \quad \uparrow -\frac{\partial H}{\partial q_i}$

$$\boxed{\frac{\partial f}{\partial t} + \{f, H\} = 0}$$

Liouville equation

in analogy w/ q_M , call $\{ _, H \} \equiv iL$

such that $\frac{\partial f}{\partial t} + iL f = 0 \Rightarrow$ formally $f(t) = e^{-iL t} f(0)$

Equilibrium means f is const in time
at every point in space, i.e.

$$\frac{df}{dt} = 0 \Rightarrow \{f, H\} = 0$$

this means f is a function of the Hamiltonian

one way of restating this is if

$$f = \mathcal{F}(H) = \sum_{n=0}^{\infty} c_n H^n \quad (\text{power series expansion})$$

$$\text{then } \{f, H\} = \left\{ \sum_{n=0}^{\infty} c_n H^n, H \right\} = \sum_{n=0}^{\infty} c_n \{H^n, H\} = 0 \quad \leftarrow \text{Exercise}$$

$\frac{df}{dt} = 0$ only determines \mathcal{F} up to a const factor

$$\text{so } f(x, t) = \frac{1}{Z} \mathcal{F}(H(x))$$

$$\text{where } Z = \int d\vec{x} \mathcal{F}(H(x))$$

This is called the partition function and
it counts the total number of microstates
accessible in the ensemble

Ensemble averaging if we knew Z (at Eq.)

$$\langle A \rangle = \int d\vec{x} A(x) \mathcal{F}(H(x)) / Z$$

The form of \mathcal{F} will depend on the ensemble

Microcanonical Ensemble

For an isolated macroscopic system with a given set of all microstates with the same macroscopic properties are equally likely
(equal - a priori probabilities postulate)

If we have N particles in a box of volume V with no exchange of energy, then it's a conservative system & dynamics will follow Ham. Eqns. of Motion

Hence at equilibrium, $\mathcal{F}(H) = \delta(H(x) - E)$

where $\delta(x)$ is the Dirac delta function, with the property $\int_{-\infty}^{\infty} dx \delta(x-a) f(x) = f(a)$
units of x^{-1}

The microcanonical partition function

$$\mathcal{R}(N, V, E) \propto \int d\vec{x} \delta(H(x) - E)$$

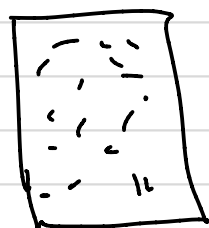
counting num points on hypersurface

The proportionality const can be set by comparison to expts, but

here we will just note it should have a $1/N!$ in it since classical particles are indistinguishable

Connection to thermodynamics

Suppose we have our system



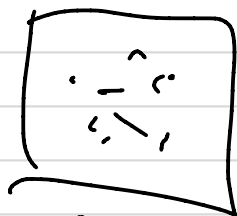
N, V, E

with $\Omega(N, V, E)$ states

If we have Z copies of the system still in isolation, how many states are there



$\Omega(N, V, E)$



$\Omega(N, V, E)$

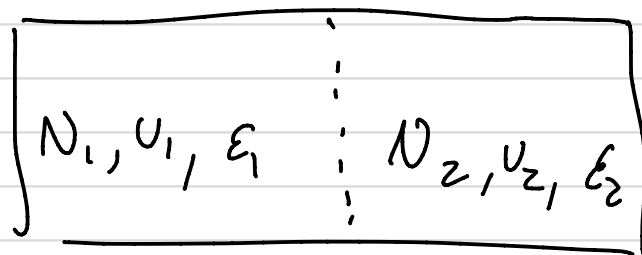
Total $\Omega(2N, 2V, 2E)$

$$= \Omega(N, V, E) \cdot \Omega(N, V, E)$$

Extensive functions double when the system size doubles, etc. What would an extensive function of Ω be?

$f(\Omega) \propto \log \Omega$ has this property

Suppose we take our two systems
 not exactly identical and let only E flow,
 whole system is isolated



$$E_2 = E_{\text{tot}} - E_1$$

Since energy can move from one side to
 the other, E_1 can be any value from 0 to
 E_{tot} . Which value is most likely?

Same as saying which value of E_1 maximizes $\Omega(E_{\text{tot}}, N_1 + N_2, U_1 + U_2)$

To max/min Ω , we do $0 = \frac{d\Omega}{dE_1}$, equivalently

$$0 = \frac{d \log \Omega}{dE_1}, \quad \text{b/c } \log \Omega \text{ monotonically increasing}$$

$$0 = \frac{d \log \Omega}{dE_1} = \frac{d}{dE_1} (\log(\Omega_1 \Omega_2)) = \left(\frac{d \log \Omega(N_1, U_1, E_1)}{dE_1} \right)_{N_1, U_1} + \left(\frac{d \log \Omega(N_2, U_2, E - E_1)}{dE_1} \right)_{N_2, U_2}$$

$$\Rightarrow \left(\frac{d \log \Omega(N_1, U_1, E_1)}{dE_1} \right)_{N_1, U_1} = - \left(\frac{d \log \Omega(N_2, U_2, E - E_1)}{dE_1} \right)_{N_2, U_2} = \left(\frac{d \log \Omega(N_2, U_2, E_2)}{dE_2} \right)_{N_2, U_2}$$

Thermodynamics says heat will flow
 from one part to the other until the
 temperatures are equal

So somehow we expect $\left(\frac{\partial \log \mathcal{R}}{\partial \epsilon}\right)_{N, U}$ to be related to the temperature

Basic thermo will tell us (next) that $\frac{1}{T} = \left(\frac{\partial S}{\partial \epsilon}\right)_{N, U}$

hence if we associate $S(N, U, \epsilon) = k_B \log \mathcal{R}(N, U, \epsilon)$

we get that ① 2 bodies in contact equalize $1/T \equiv \beta$

② Entropy is maximized for a closed system at equilibrium

This connects microscopic states to one macro observable (& others, see next)