

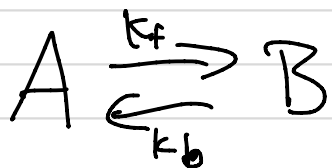
Lecture 23:

Non-Equilibrium Pt 3

Other kinds of Brownian Motion

The general theory we are learning is useful for other kinds of random processes besides a particle in solution. To illustrate this, let's look @ chemical reactions

Simplest reaction is



expect $\frac{dA}{dt} = Bk_b - Ak_f$

$$\frac{dB}{dt} = Ak_f - Bk_b$$

In this case, we know

$$A + B = N \quad \leftarrow \text{const \# molecules}$$

$$\text{@ } E_{eq} \text{ still true, } A_{eq} + B_{eq} = N$$

In the spirit of our prev work, think about

$$A = A_{eq} + C, \quad A \text{ at a particular time is a deviation from eq}$$

$$\text{This means } B = B_{eq} - C$$

C measures reaction condition from all in

$$A \quad (C = B_{eq}) \text{ to all in } B \quad (C = -A_{eq})$$

Lastly, we have our detailed balance condition

$$A_{eq} k_f = k_b B_{eq}$$

Combining this info, we have

$$\frac{d(A_{eq} + C)}{dt} = -k_f(A_{eq} + C) + k_b(B_{eq} - C)$$

$$\frac{d(B_{eq} - C)}{dt} = k_f(A_{eq} + C) - k_b(B_{eq} - C)$$

$$\frac{dA_{eq}}{dt} = \frac{dB_{eq}}{dt} = 0$$

$$\text{Subtract: } 2 \frac{dC}{dt} = 2k_b(B_{eq} - C) - 2k_f(A_{eq} + C)$$

$$\Rightarrow \frac{dC}{dt} = -(k_f + k_b)C$$

$$\Rightarrow C(t) = e^{-(k_f + k_b)t} \leftarrow \tau_{rxn} = \frac{1}{k_f + k_b}$$

Macroscopic diff from eq decays to eq. exponentially

Ozuges Regression Hypothesis (1931)

Small fluctuations decay on the average

@ eq the same way as macroscopic ^{non eq} deviations

[not really a hypothesis, more like so far always true theory/law]

Makes sense, how would you know whether prepared in this state, or result of true dynamics

$$\Rightarrow \langle C(t)C(t') \rangle = \langle C^2 \rangle_{eq} e^{-(k_1 + k_2)(t-t')}$$

However, it can't be true that

the non eq condition goes to $C=0, \dots$

... and then the number of A & B are fixed, they have to fluctuate randomly

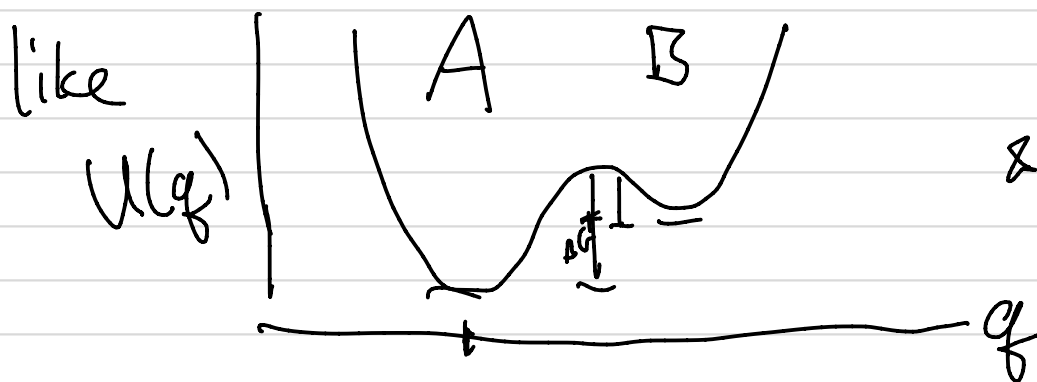
Have to maintain a $\langle C^2 \rangle_{eq}$ that is non-zero
 this is just like Brownian motion

so, postulate $\frac{dC}{dt} = -(k_1 + k_2)C + \delta F$

\rightarrow & now $\langle \delta F(t) \delta F(t') \rangle = 2(k_1 + k_2) \langle C^2 \rangle_{eq} \delta(t - t')$

HW

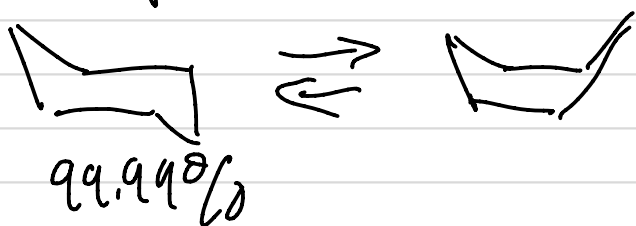
This is a macroscopic view of chemical eq,
 but where do these rate constants come from.
 For this simple prob, we expect something



& $P_A / P_B \approx \Delta G$
 & $k_{A \rightarrow B} \propto e^{-\beta \Delta G_A^\ddagger}$
 & $k_{B \rightarrow A} \propto e^{-\beta \Delta G_B^\ddagger}$

Now we have to connect this to the microscopic stat mech theory we've learned all semester. Molecularly, we still have $Q(x,p) = C \int d\vec{x} \int d\vec{p} e^{-\beta H(\vec{x}, \vec{p})}$

Real problem example could be



Define q by a collective coordinate

transition state @ q^\ddagger

Can define a function $H_A(q) = \begin{cases} 1, & q < q^\ddagger \\ 0, & q \geq q^\ddagger \end{cases}$ (can I in A)

$$\langle H_A \rangle = X_A = \frac{A_{eq}}{A_{eq} + B_{eq}}$$

Since H_A is 1 in A & 0 otherwise

$$\langle H_A^2 \rangle = \frac{H_A(A)^2 P(A) + H_A(B)^2 P(B)}{P(A) + P(B)} = \frac{P(A)}{P(A) + P(B)} = X_A$$

$$\Rightarrow \langle \delta H_A^2 \rangle = \langle H_A^2 \rangle - \langle H_A \rangle^2 = X_A - X_A^2 \\ = X_A(1 - X_A) = X_A X_B$$

Now, already said

$$\langle C(t)C(0) \rangle = e^{-t/\tau_{rxn}} \cdot \langle C^2 \rangle$$

similarly at a microscopic level

$$\langle \delta H_A(q(t)) \delta H_A(q(0)) \rangle = e^{-t/\tau_{rxn}} \langle \delta H_A^2 \rangle$$

$$\Rightarrow e^{-t/\tau_{rxn}} = \frac{\langle \delta H_A(q(t)) \delta H_A(q(0)) \rangle}{X_A X_B}$$

call $H_A(q(t)) = H_A(t)$ for simplicity

since we care how the number in A is changing in time

take time deriv

$$-\frac{1}{\tau_{rxn}} e^{-t/\tau_{rxn}} = \frac{\langle \dot{\delta H}_A(t) \delta H_A(0) \rangle}{X_A X_B}$$

Last time sort of discussed

↙ @ eq 50

HW?

$$\langle A(t)A(t') \rangle = \langle A(0)A(t-t') \rangle = \langle A(t-t')A(0) \rangle \\ \Rightarrow \langle A(0)A(t) \rangle = \langle \dot{A}(0)A(t) \rangle$$

For our case $-\langle \delta H_A(0) \delta H_A(t) \rangle = \langle \delta H_A(0) \delta H_A(t) \rangle$

Important to note $\frac{dH[q]}{dt} = \dot{q} \frac{d}{dq} H_A = -\dot{q} \delta(q - q^*)$

B changes from $0 \rightarrow -\infty \rightarrow 0$ instantaneously

$$\begin{aligned} \text{So } -\langle \delta H_A(0) \delta H_A(t) \rangle &= \langle -\dot{q}(0) \delta(q(0) - q^*) \delta H_A(q(t)) \rangle \\ &= \langle \dot{q}(0) \delta(q(0) - q^*) \delta H_B(q(t)) \rangle \end{aligned}$$

Since $H_B = 1 - H_A$

$$\text{and } \langle \dot{q}(0) \delta(q(0) - q^*) \rangle = 0$$

b/c velocity & configs are uncorrelated

Finally

$$\frac{1}{\tau_{rxn}} e^{-t/\tau_{rxn}} = \frac{1}{\lambda_A \lambda_B} \langle \dot{q}(0) \delta(q(0) - q^*) H_B(q(t)) \rangle$$

$\dot{q}(0) \delta(q(0) - q^*)$ on surface

Right side is flux crossing surface if ends up in B

Left side is simple exponential & cont account for really short time flux, here

evidence regression hypothesis is true only after coarse-graining over short time scales, so expect this to be true for $\tau_{mol} \ll t \ll \tau_{rxn}$ (fast barrier crossing)

$$\text{If so } \frac{1}{\tau_{rxn}} = k_f + k_b = \frac{1}{\chi_A \chi_B} \langle V(\omega) \delta(q - q^*) H_B(q(t)) \rangle$$

mult by χ_B & detailed balance

$$\begin{aligned} \chi_B(k_f + k_b) &= \frac{B}{A+B} (k_f + k_b) = \frac{B/A}{1+B/A} (k_f + k_b) = \frac{\frac{k_f}{k_b}}{1 + \frac{k_f}{k_b}} (k_f + k_b) \\ &= k_f \end{aligned}$$

$$\text{So } k_f = \frac{1}{\chi_A} \langle V(\omega) \delta(q - q^*) H_B[q(t)] \rangle$$

This connect microscopic behavior at the transition state to the macroscopic reaction rate

