

# Homework 2: Microcanonical ensemble

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Due: Oct 1, 2021 at 5pm

1. *The Gamma ( $\Gamma$ ) function as a generalized factorial.* The  $\Gamma$  function has an important property that we will use to derive the microcanonical partition function for an ideal gas, which is that it acts as a factorial operator for integers. The gamma function is defined as

$$\Gamma(z) = \int_0^{\infty} x^{z-1} e^{-x} dx \quad (1)$$

Show that  $\Gamma(N + 1) = N! = N(N - 1)(N - 2)\dots(2)(1)$  by the following steps:

- (a) Show that  $\Gamma(1) = 1$
  - (b) Using integration by parts, show that  $\Gamma(z + 1) = z\Gamma(z)$  for  $z$  as a real number
  - (c) Using the above two properties, show that  $\Gamma(N + 1) = N!$  for  $N$  integers greater or equal to 1.
2. *Surface area of an  $N$ -dimensional sphere.* In order to derive the partition function for an ideal gas of particles, we needed to know the formula for the surface area of a sphere in  $3N$ -dimensions. Let's actually derive this formula so we can see where it comes from. It takes advantage of knowledge and similar techniques you learned from Homework 1.

First, a quick definition, the way mathematicians write things. The volume of a sphere-like object of radius 1 (every point is distance  $r \leq 1$  from the origin) in  $d$  dimensions is called  $V_d$ . Confusingly, the surface area of that same object is called  $S_{d-1}$  (since it is an object which is 1-dimension "flatter"). The volume of  $d$ -sphere of radius  $R$  is  $V_d R^d$  and the surface area is  $S_{d-1} R^{d-1}$ . So, e.g.  $V_3 = \frac{4}{3}\pi R^3$  and  $S_2 = 4\pi R^2$ .

- (a) A  $d$ -sphere can be built by adding up a bunch of shells of smaller radius (Think, make up a disk ( $d = 2$ ) by drawing a bunch of concentric circles). The

volume is the addition of all of the surface areas of the shells. So,

$$V_d = \int_0^1 dr S_{d-1} r^{d-1} = S_{d-1} \frac{1}{d} r^d \Big|_0^1 = \frac{S_{d-1}}{d} \quad (2)$$

Show that this formula is correct for  $d = 2$  and  $d = 3$  using your knowledge of circles and spheres.

- (b) We already used polar and spherical coordinates to derive certain things. We saw that if the function we want to integrate only depends on distance from the origin, and  $r^2 = \sum_{i=1}^d x_i^2$ , then

$$\int dx_1 dx_2 \dots dx_d f(r) = \int_0^\infty dr S_{d-1} f(r) r^{d-1} = S_{d-1} \int_0^\infty dr f(r) r^{d-1}, \quad (3)$$

(do you see how this is true for spherical and polar coordinates?). If  $f(r) = 1$  then we get back Eq. 2 (with different integration limits). If we can solve both sides of this integral for any  $f(r)$ , then we will have a general formula for  $S_{d-1}$ .

Do the following steps:

- i. Similar to last week, define  $I = \int_{-\infty}^\infty e^{-x^2} dx$ . Write  $I^d$  as a product of integrals over different coordinate variables and get an integral over a function of  $r$  that looks like the left hand side of Eq. 3.
- ii. Now that you have  $f(r)$ , show that the right-hand side can be rewritten as something proportional to  $\Gamma(d/2)$ . You will have to do a substitution.
- iii. Since we know the value of  $I$  from last time, we also know the value of  $I^d$ . Equate this value with the formula from the preceding step to get the final result for surface areas, (book equation 3.5.14, with  $n = d - 1$ )

$$S_{d-1} = 2 \frac{\pi^{d/2}}{\Gamma(d/2)} \quad (4)$$

- iv. Given this result, what is the formula for the Volume of a  $d$ -sphere of radius 1,  $V_d$ ? Use the fact that  $\frac{d}{2} \Gamma(\frac{d}{2}) = \Gamma(\frac{d}{2} + 1)$  to simplify the equation.
- (c)
- i. Use the definition of the gamma function to show  $\Gamma(1/2) = \sqrt{\pi}$ . Hint: The right hand side should look familiar here. Reverse your substitution from earlier to get a familiar integral.
  - ii. Use this result and the formula for  $V_d$  above to show that  $V_3$  has the value you expect for a sphere.

3. *Microcanonical ideal gas, finishing the derivation.* In class, we worked out that the

microcanonical partition function has the form

$$\Omega(N, V, E) = \frac{E_0 V^N}{h^{3N} N!} \int_{-\infty}^{\infty} dp_1 dp_2 \dots dp_{3N} \delta \left( \sum_{i=1}^{3N} \frac{p_i^2}{2m} - E \right) \quad (5)$$

If we substitute  $p_i = \sqrt{2m}y_i$  here, we get

$$\Omega(N, V, E) = \frac{E_0 V^N (2m)^{3N/2}}{h^{3N} N!} \int_{-\infty}^{\infty} dy_1 dy_2 \dots dy_{3N} \delta \left( \sum_{i=1}^{3N} y_i^2 - E \right) \quad (6)$$

(a) Use Eq. 3 and Eq. 4 to rewrite this as

$$\Omega(N, V, E) = \frac{E_0 V^N (2m)^{3N/2}}{h^{3N} N!} \frac{2\pi^{3N/2}}{\Gamma(3N/2)} \int_0^{\infty} dr \delta(r^2 - E) r^{3N-1} \quad (7)$$

(b) Use the formula from class to split this delta function into two delta functions, then perform the integral to show that

$$\Omega(N, V, E) = \frac{E_0 V^N (2m)^{3N/2}}{h^{3N} N!} \frac{\pi^{3N/2}}{\Gamma(3N/2)} \frac{E^{3N/2}}{E} \int_0^{\infty} dr \delta(r^2 - E) r^{3N-1} \quad (8)$$

$$= \frac{E_0}{E} \frac{1}{N!} \frac{1}{\Gamma(3N/2)} \left[ V \left( \frac{2\pi m E}{h^2} \right)^{3/2} \right]^N \quad (9)$$

(c) *Stirling's approximation* says that  $\log(N!) \approx N \log N - N$  or equivalently  $N! \approx N^N e^{-N}$  for large  $N$ . Substituting  $\Gamma(X - 1) = X!$ , using the approximation  $N - 1 \approx N$  for very large  $N$ , and using *Stirling's approximation* for this term, show that

$$\Omega(N, V, E) = \frac{E_0}{N!} \left[ V \left( \frac{4\pi m E e}{3N h^2} \right)^{3/2} \right]^N \quad (10)$$

(d) Using our formula for entropy,  $S = k_B \log(\Omega(N, V, E))$ , substituting the formula we derived  $E = \frac{3}{2} N k_B T$ , and neglecting  $k_B \log(E_0)$  both because it is an arbitrary constant and because it is not proportional to  $N$ ,

i. Show that the entropy of a monatomic ideal gas is:

$$S(N, V, E) = N k_B \log \left[ V \left( \frac{2\pi m k_B T}{h^2} \right)^{3/2} \right] + \frac{3N k_B}{2} - k_B \log(N!) \quad (11)$$

ii. And using *Stirling's approximation*, derive the *Sackur-Tetrode equation*:

$$S(N, V, E) \approx N k_B \log \left[ \frac{V}{N} \left( \frac{2\pi m k_B T}{h^2} \right)^{3/2} \right] + \frac{5}{2} N k_B \quad (12)$$

4. *Gibbs Paradox*. Show using the entropy formula from above (Eq. 11), that the entropy of mixing of two boxes of identical particles is non-zero unless the  $1/N!$  factor is included in the microcanonical partition function (follow Tuckerman book Section 3.5.1).